

# STRUCTURE AND ELECTRONIC PROPERTIES OF Ni NANOPARTICLES DEPOSITED ON CeO<sub>2</sub> THIN FILMS

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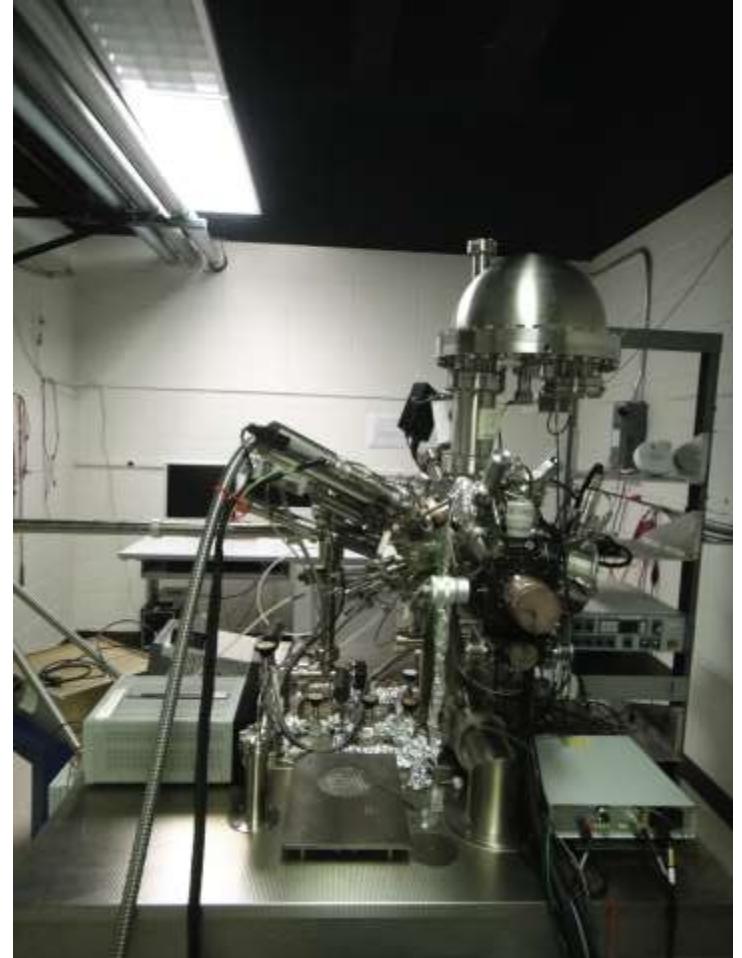
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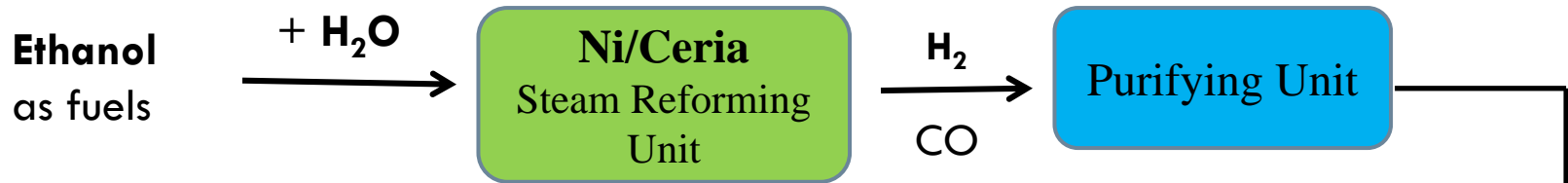
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# Overview

- Introduction
- Experimental Approach
- X-Ray Photoelectron Spectroscopy (XPS) Results
- Scanning Tunneling Microscopy (STM) Results



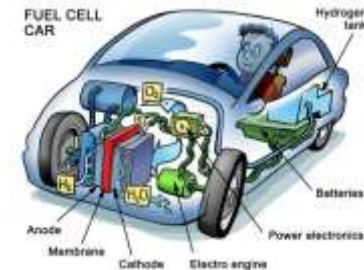
# Ni with Ceria Support Creates Ecofriendly Energy Resources



- Ni is a good catalyst
  - ▣ Inexpensive
  - ▣ Readily available
  - ▣ Highly Active
- But the system is not well understood



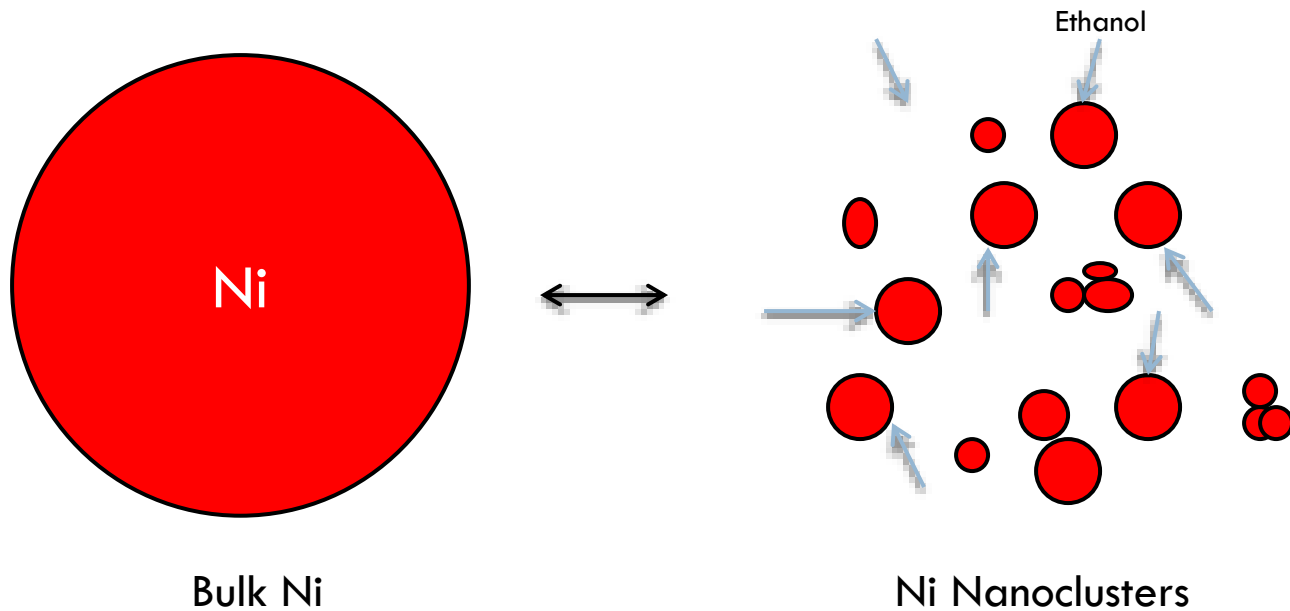
NASA uses hydrogen fuel to launch the space shuttles. Credit: NASA



+ others

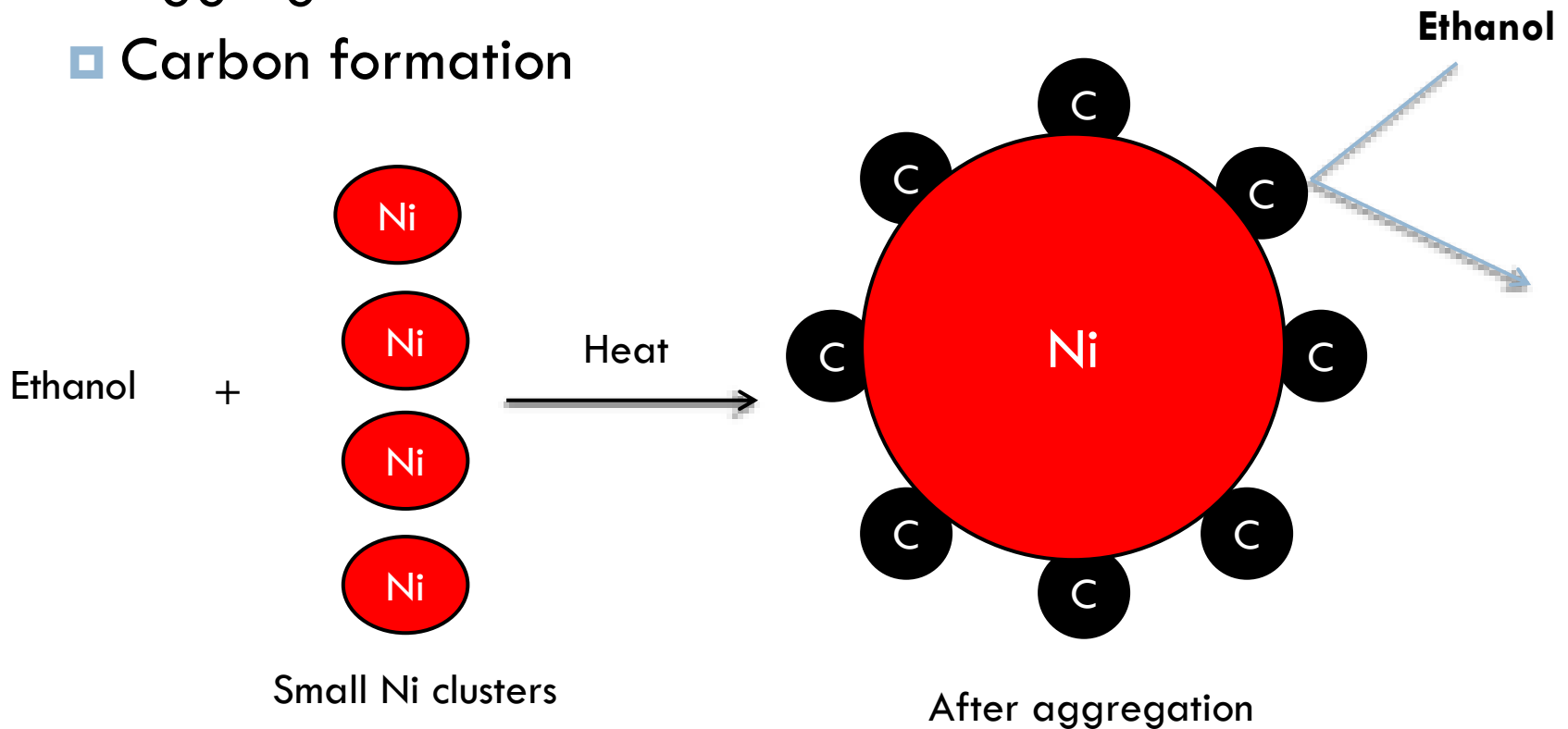
# Size and Structure Do Matter

- Ni nanoparticle can increase the reactivity
  - ▣ Larger surface areas = Higher reactivity
- Structures of Ni can affect the reactivity



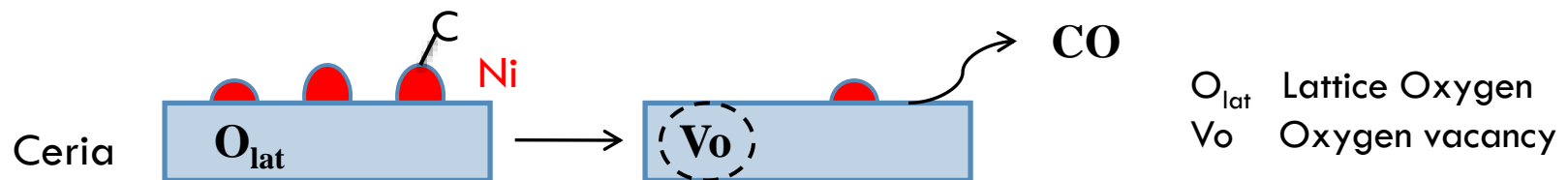
# Problems of Using Pure Ni as the Catalyst

- Ni becomes deactivated
  - ▣ Aggregation
  - ▣ Carbon formation



# Ceria Support Provides a Potential Solution

- Ceria is reducible:  $\text{Ce}^{4+} \leftrightarrow \text{Ce}^{3+}$
- Ceria:  $\text{CeO}_2$ ,  $\text{Ce}_2\text{O}_3$ , mixture of the two
- Oxidized Ceria:  $\text{CeO}_2$
- Reduced Ceria:  $\text{CeO}_x$  ( $2 < x < 1.5$ )
- Ceria can eliminate carbon deposition on Ni



- Ceria can inhibit aggregation of Ni



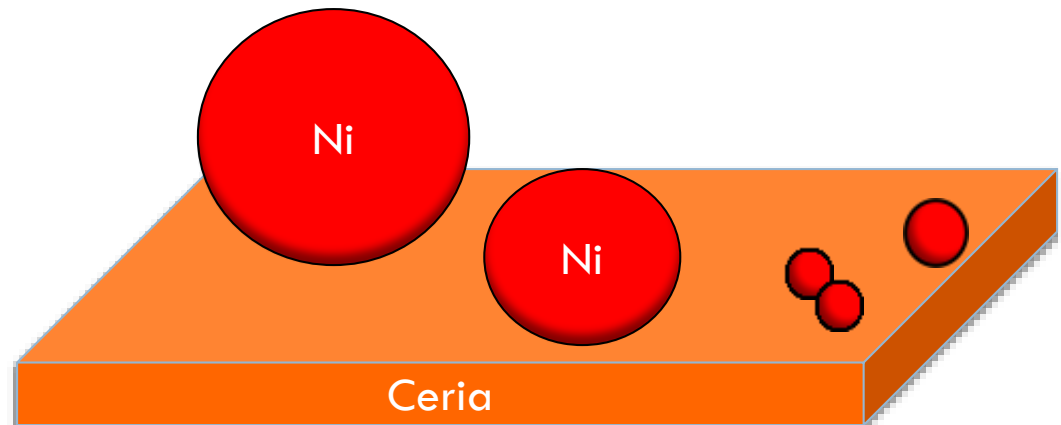
# Experimental Approach

1. Preparation  
of Ceria  
Supported of Ni

2. Structure and  
Electronic  
Characterization

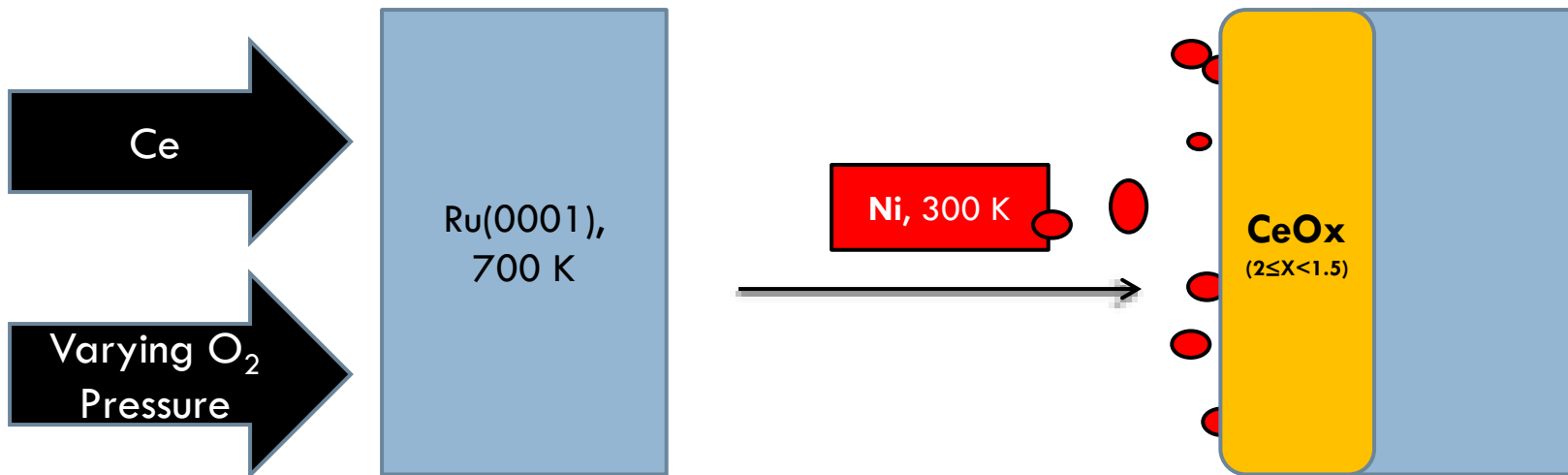
3. Ethanol  
Chemistry

- Ni catalytic reactivity can be affected by
  - Redox properties of ceria support ( $\text{Ce}^{4+}$ ,  $\text{Ce}^{3+}$ )
  - Ni and ceria interaction
  - Size
  - Structure



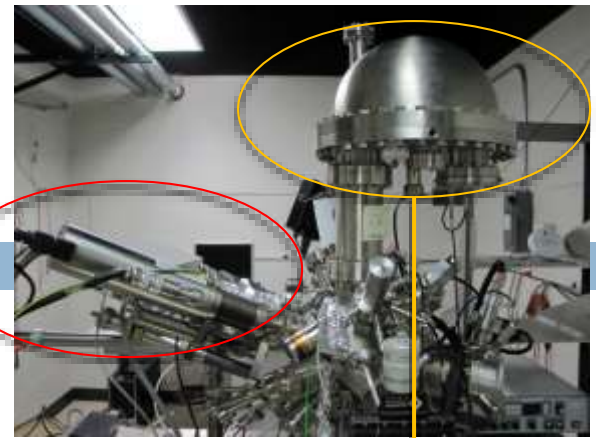
# Preparation of Ceria and Deposition of Ni particles

- Two types of ceria:
  - Oxidized ceria,  $\text{CeO}_2$
  - Reduced ceria,  $\text{CeO}_x$
- Ultra high Vacuum Condition: pressure:  $10^{-11}$  Torr





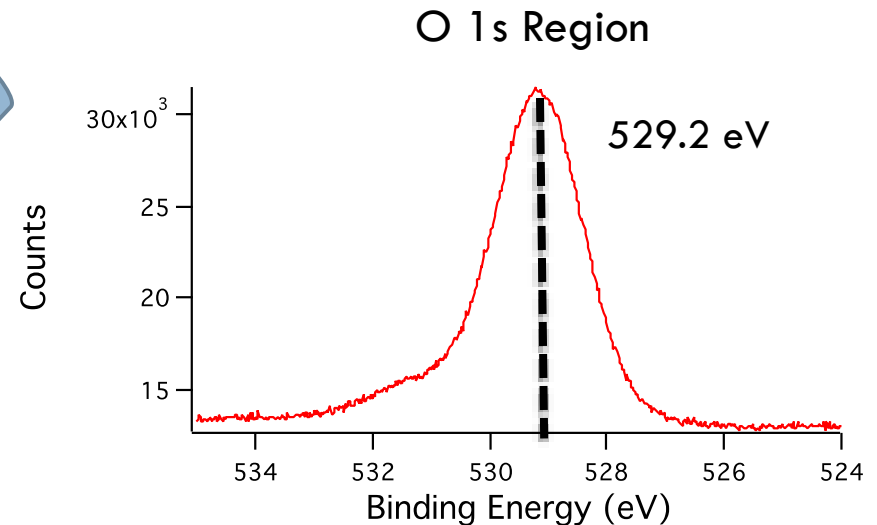
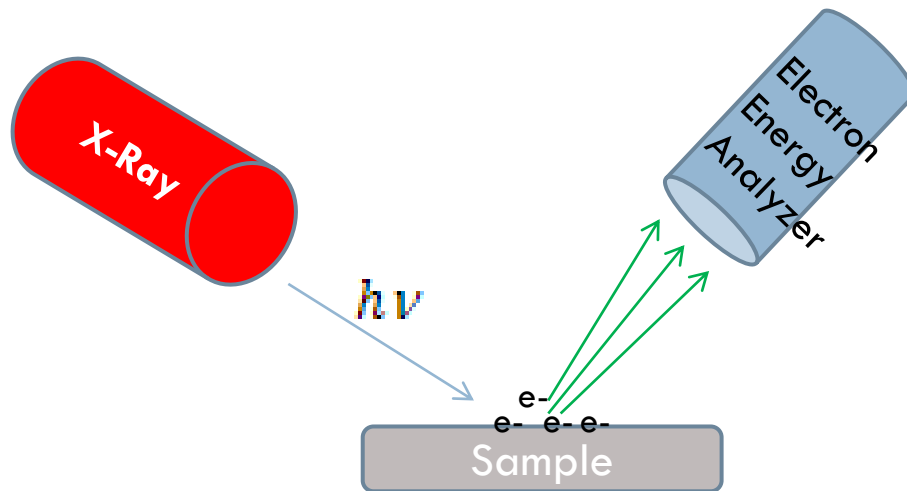
# Basic Principle of XPS



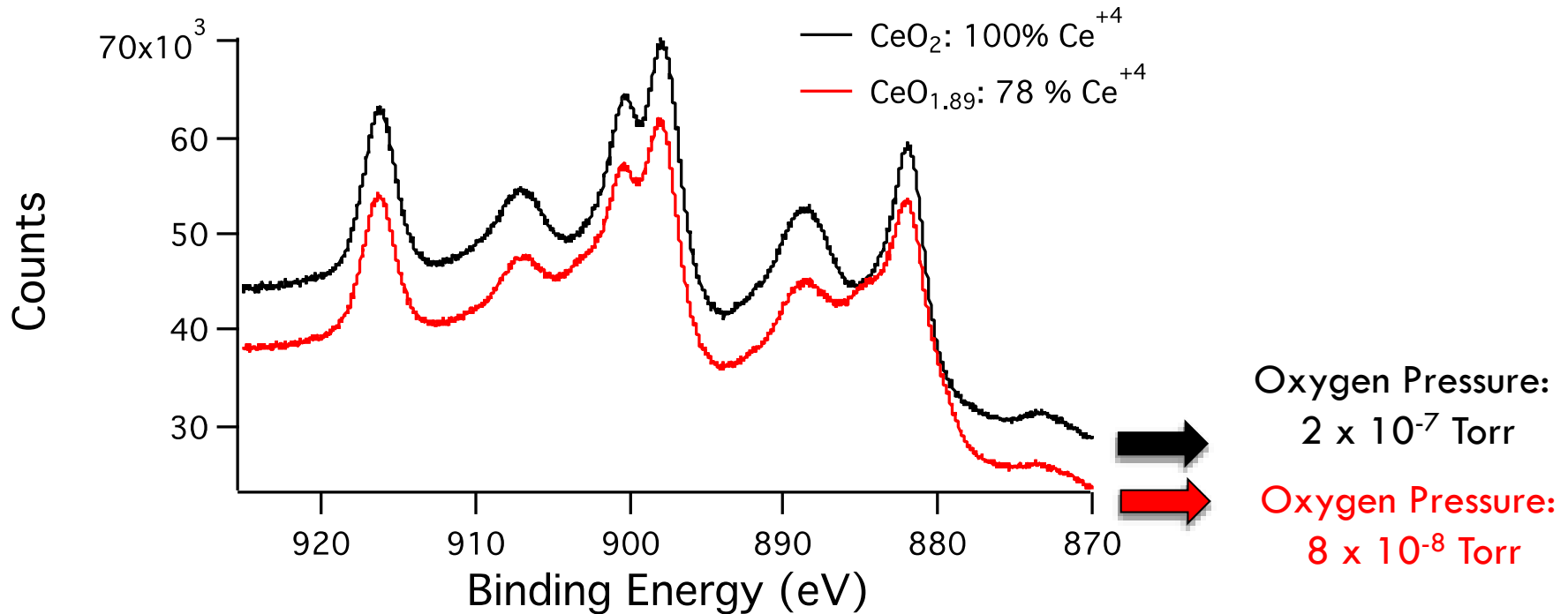
X-Ray

Electron Energy Analyzer

- Utilizes photoelectric effect
- Binding Energy = Energy of X-Ray ( $h\nu$ ) – Kinetic Energy (KE)
- Provides “finger print “ identity and the oxidation state of a sample

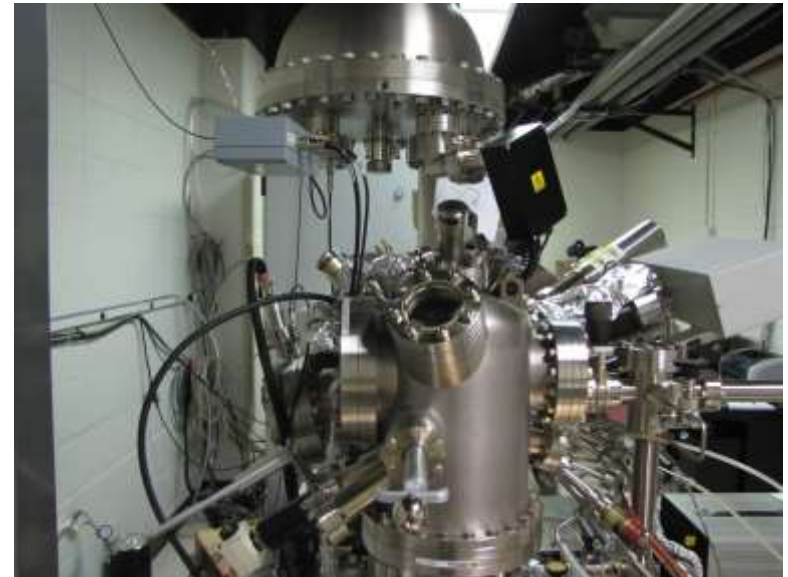
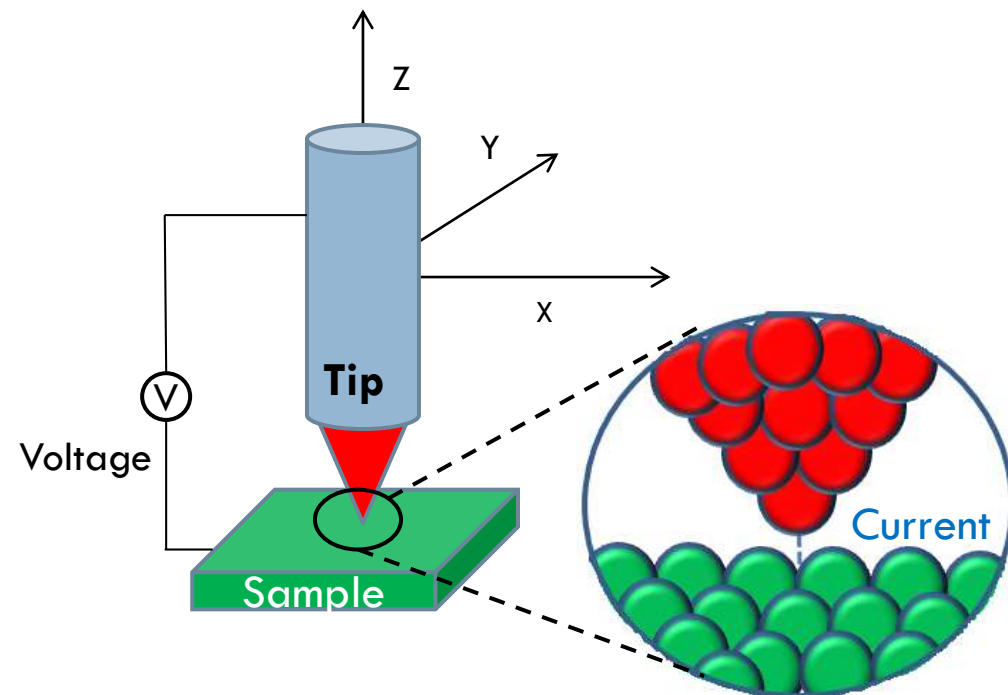


# X-Ray Photoelectron Spectroscopy of Ce 3d



# Basic Principle of Scanning Tunneling Microscopy

- Requires a sharp metal probe, a conductive sample, voltage supply, and a constant feedback loop
- Generate atomic resolution images of the surface

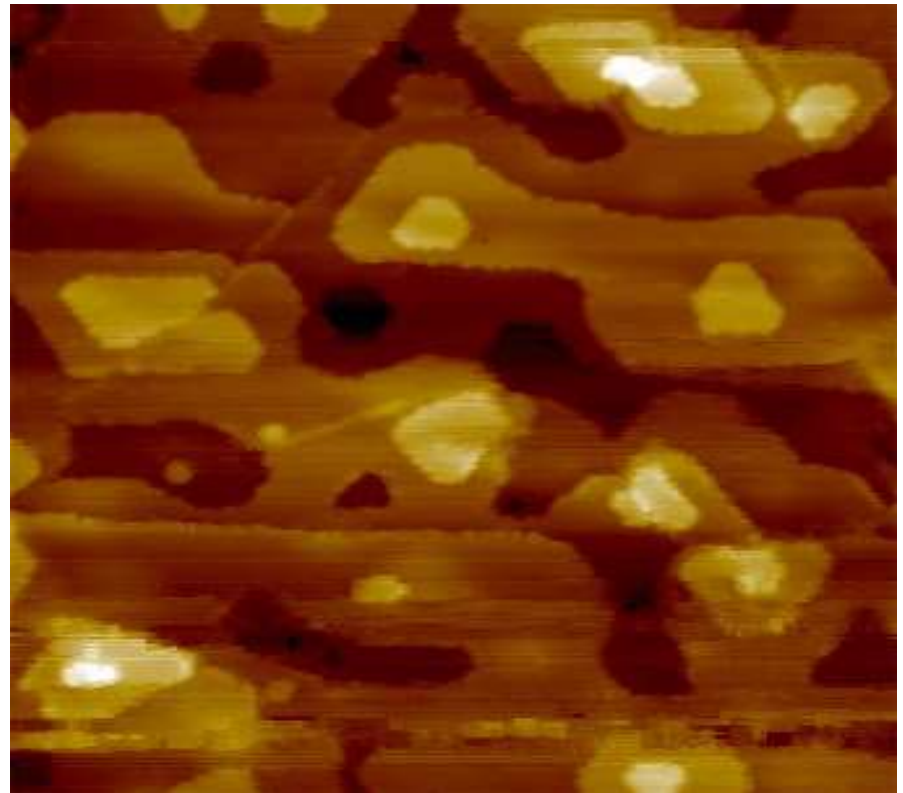


# STM Results

## □ Characteristic of Ceria:

- Flat terraces
- Layers

CeO<sub>2</sub> at 300 K

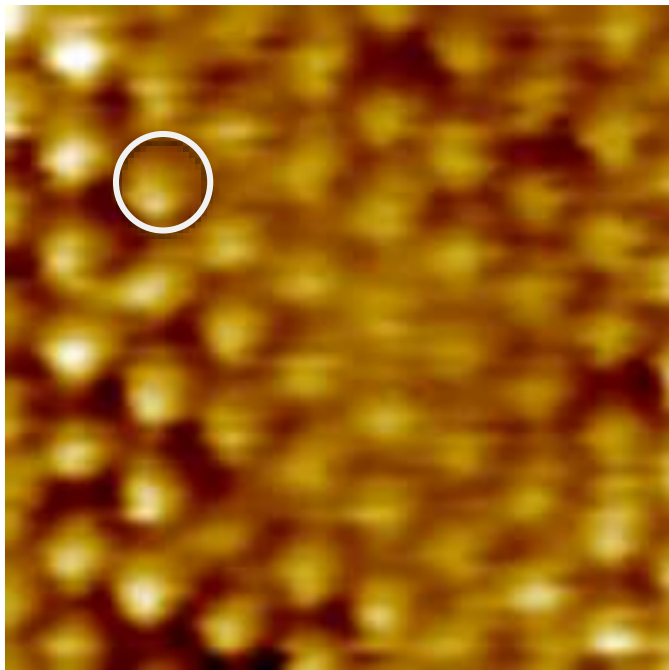


100 x 100 nm<sup>2</sup>

# Ceria at the Atomic Scale

- Ce atoms are observed
- More oxygen vacancies on  $\text{CeO}_{1.88}$

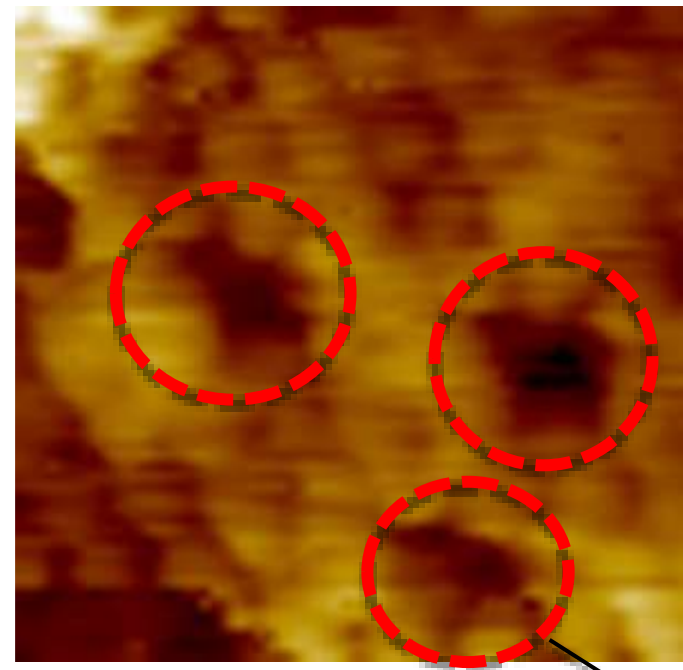
Oxidized Ceria ( $\text{CeO}_2$ )



Individual  
Ce atom

3 nm x 3 nm

Reduced Ceria ( $\text{CeO}_{1.88}$ )



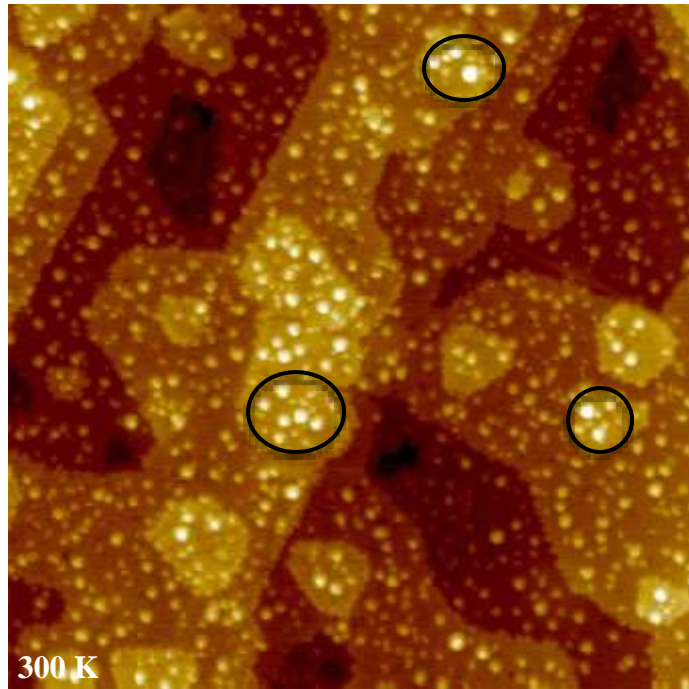
3 nm x 3 nm

Oxygen vacancies

# Ni Deposited on Ce

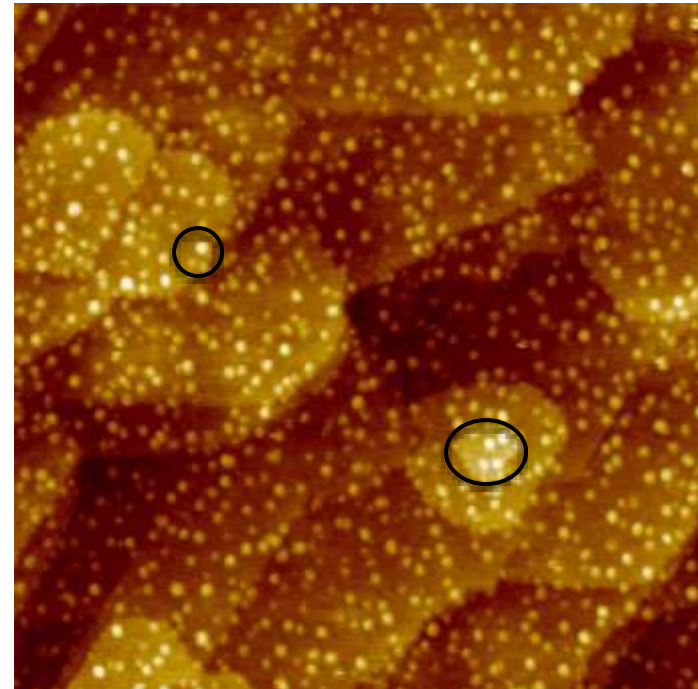
- Ni nanoparticles can form
- Ni can form nanoclusters

Ni/CeO<sub>2</sub>



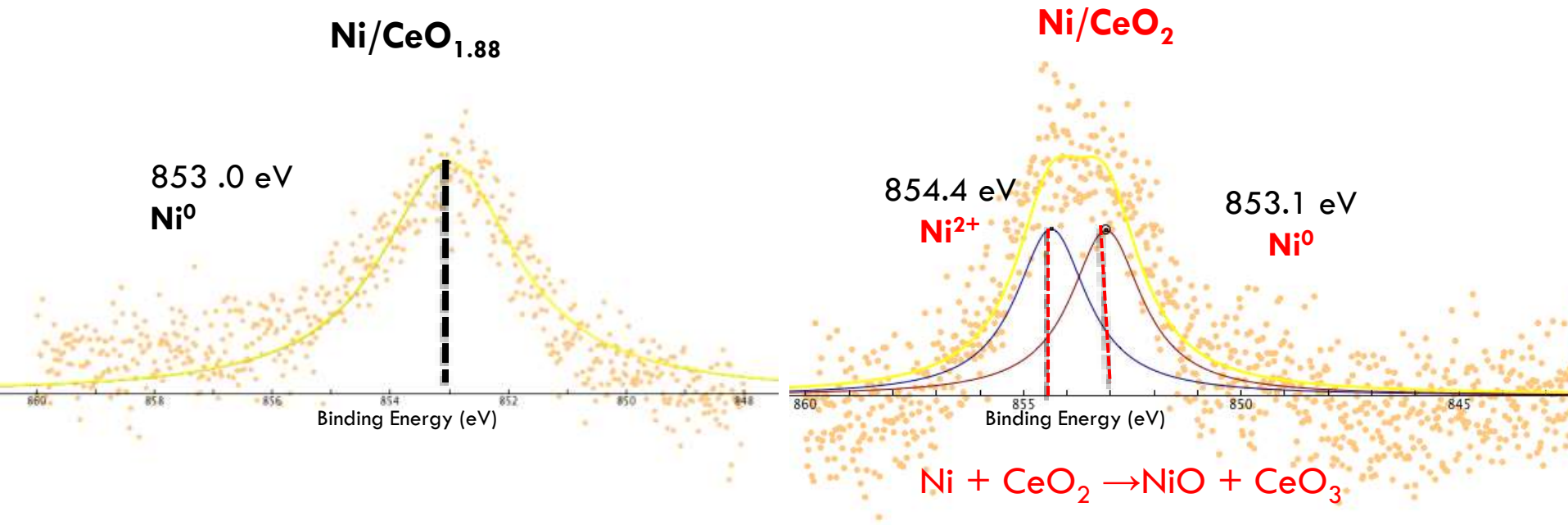
100 x 100 nm<sup>2</sup>

Ni/CeO<sub>1.88</sub>



100 x 100 nm<sup>2</sup>

# Ni 2p XPS Spectra



- Metallic Ni can form on the reduced ceria
- NiO and metallic Ni can form on the oxidized ceria

# Conclusions

**1. Preparation of  
Ceia and  
Deposition of Ni**

**2. Structure and  
Electronic  
Characterization**

**3. Ethanol  
Chemistry**

- Different Ce thin films can be prepared
- Ni nanoclusters can form
- Reduced ceria have more oxygen vacancies
- Ni remains metallic state on the reduced ceria
- NiO can form on the oxidized ceria.



# Acknowledgements

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