

# **ENHANCING DEVICE DURABILITY OF PROTON EXCHANGE MEMBRANE FUEL CELLS VIA POLYDOPAMINE COATING**

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# WHY PROTON EXCHANGE MEMBRANE (PEM) FUEL CELL?

Clean energy

Easily obtainable hydrogen and oxygen/air fuel

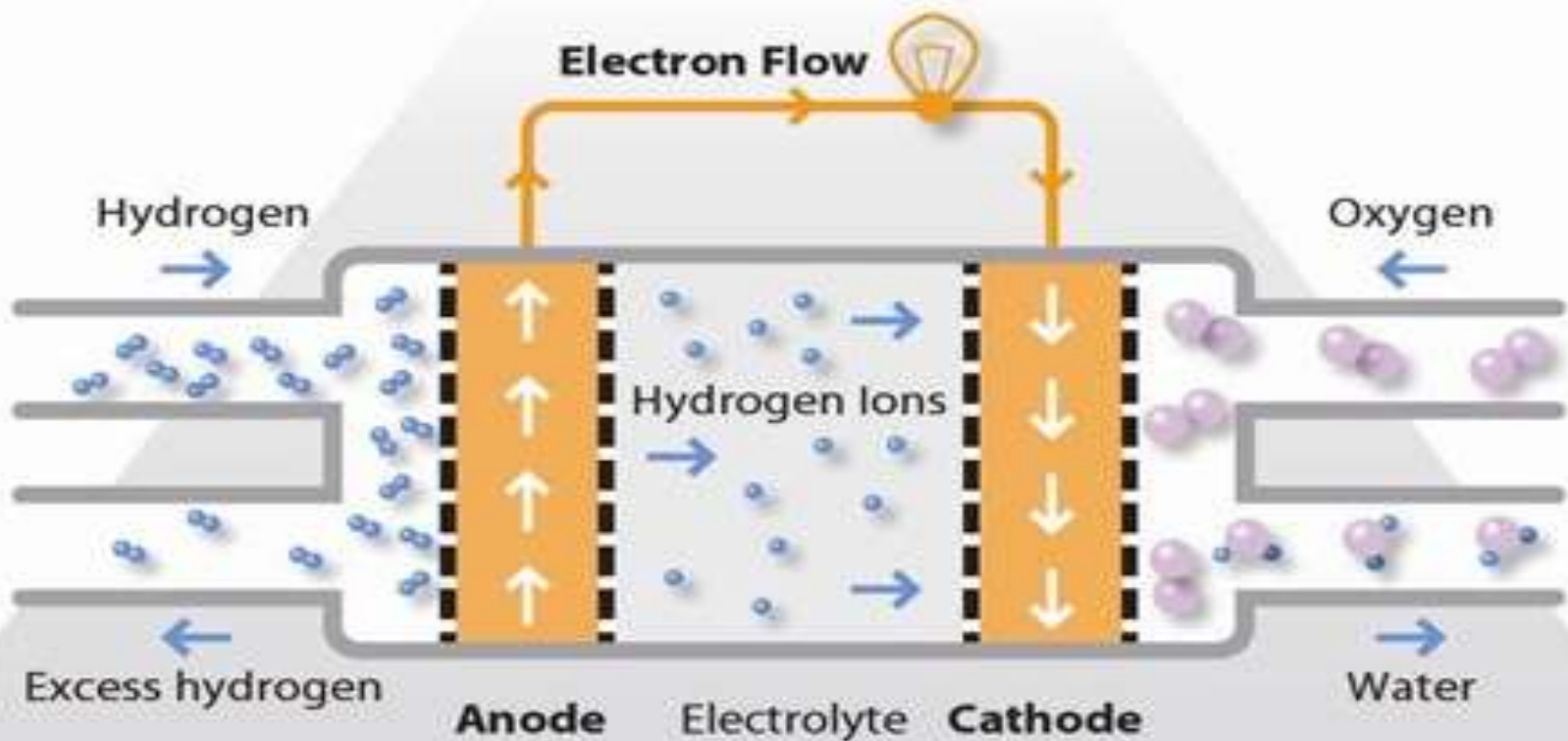
Water is the only byproduct

Multiple uses

- Large scale industry
- Small scale mobile, like automobiles



# PROTON EXCHANGE MEMBRANE (PEM) FUEL CELLS



## Key components:

Anode and Cathode

Gas Diffusion Layer (GDL), carbon paper

Proton conductive membrane, nafion 212

Precious metal catalyst, platinum

Membrane Electrode Assembly (MEA)

2 GDLs, catalyst, and membrane

# PEM FUEL CELL PROBLEMS

## High cost

- Precious metal catalysts, most commonly platinum on carbon black
  - High catalyst loading increases performance AND cost

## Low durability

- Low catalyst loading causes amalgamation due to increased surface energy
  - Catalyst amalgamation decreases performance by decreasing active catalyst sites
- Fuel crossover degrades membrane
  - This causes increased crossover and further degradation

# WHY DOPAMINE?

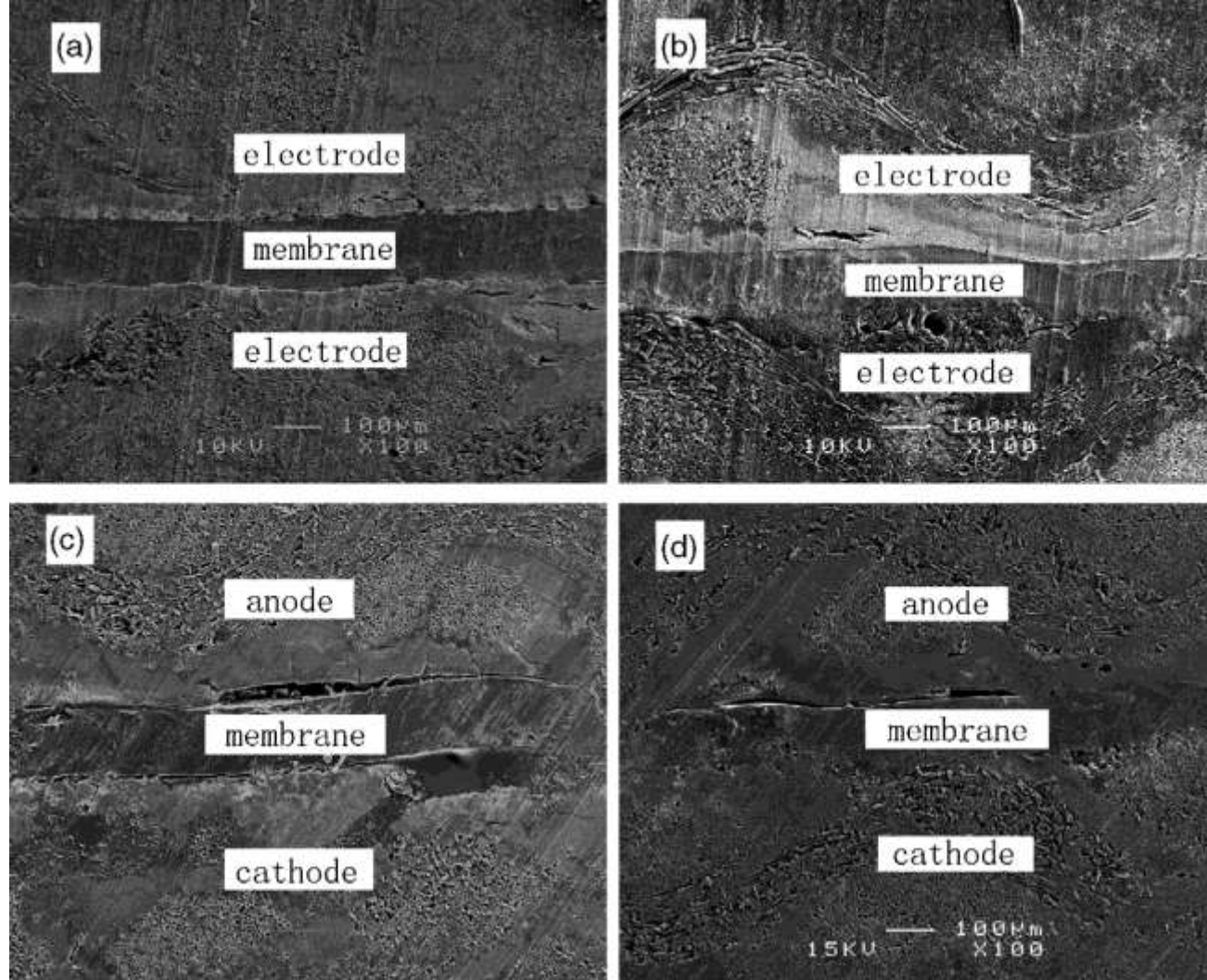
## Increased Durability

- The dopamine acts as an adhesive layer “glue” binding the MEA and decreasing mechanical stress
- Blocked pores decrease fuel crossover
  - Catechol functional groups on dopamine still allow proton permeability
  - Dopamine blocks access to the membrane backbone, lessening the effects of fuel crossover

## Decreased Cost

- The “glue” binds catalyst in place, decreasing amalgamation
  - particularly useful with lower catalyst loading





**Figure 1:** Scanning electron microscopy images (a) and (c) show the new and after lifetime test pictures of a standard MEA, note the large crack. Images (b) and (d) show before and after images for a dopamine treated MEA. The delamination is noticeably less in dopamine treated MEA.

# MEA FABRICATION PROCEDURE

## Membrane hydration and Dopamine Treatment





# Ultrasonic spray catalyst application





Heat pressing at 135C and 80psi – causes temperature damage, binds MEA



Handwritten notes on a sticky note, partially legible, mentioning "fuel cell" and "test system".



FRAGILE  
Scribner Associates

# MEA TESTING

Linear sweep voltammetry (LSV) tests

Cyclic voltammetry (CV) tests

Performance tests

Long term durability test

Pre-break in

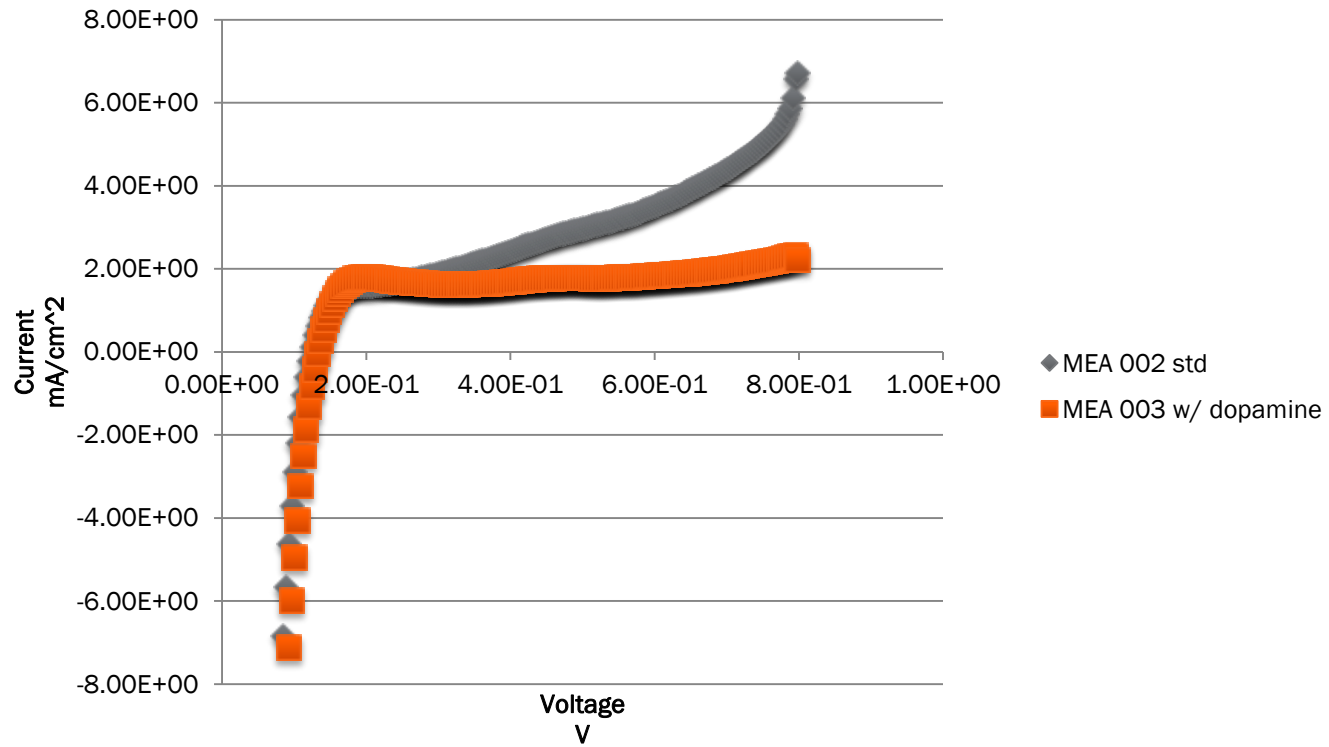
Standard tests

High temp tests

- High breakdown

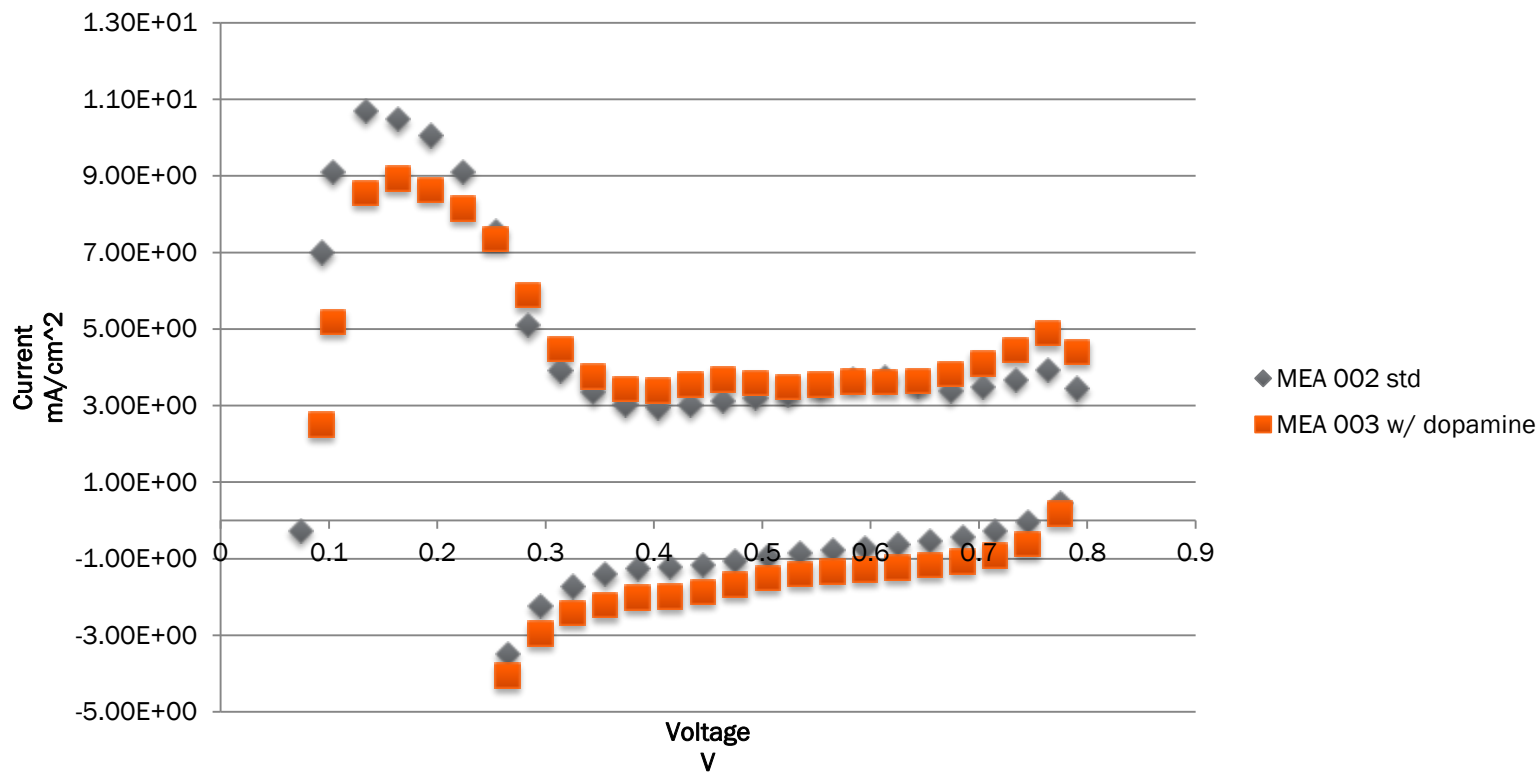
# LSV RESULTS

## LSV at 120/90/90C



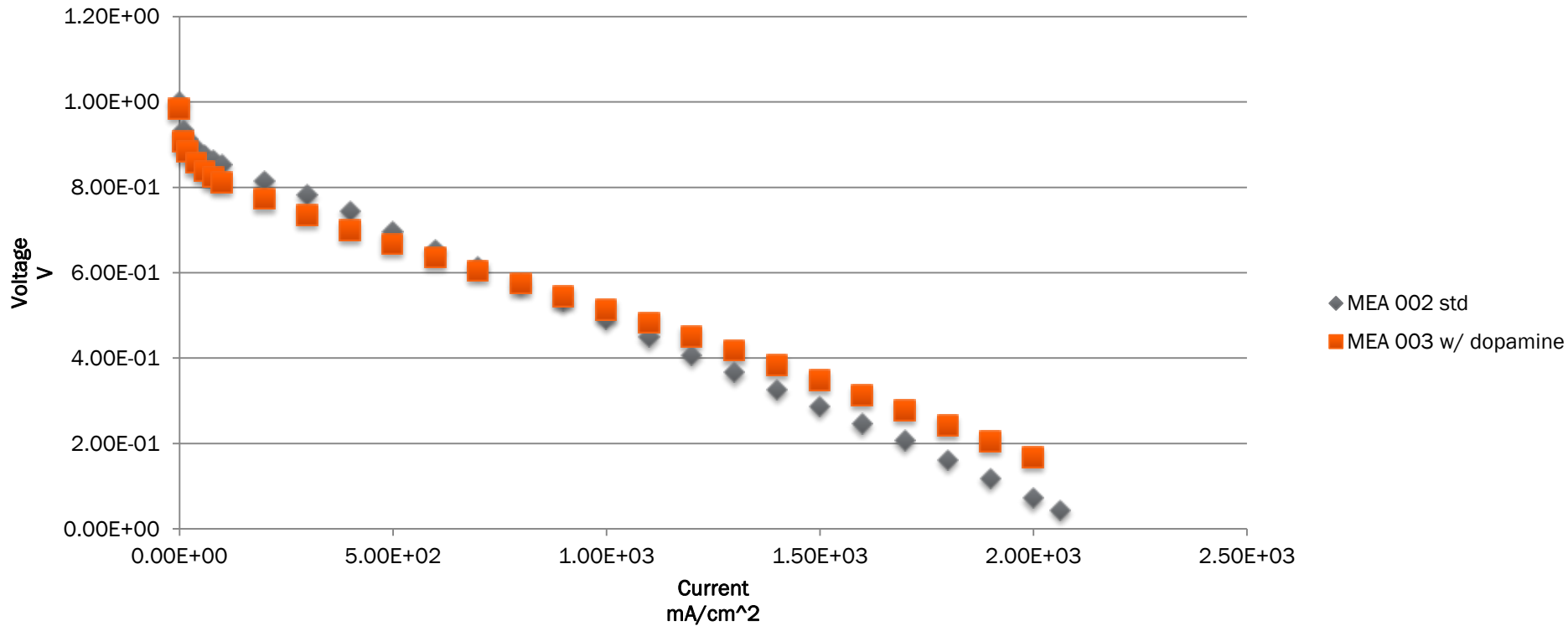
# CV RESULTS

## CV at 25/25/25C postbreak-in



# PERFORMANCE CURVE RESULTS

Performance Curve at 80/80/73C 1.5atm backpressure



OCV

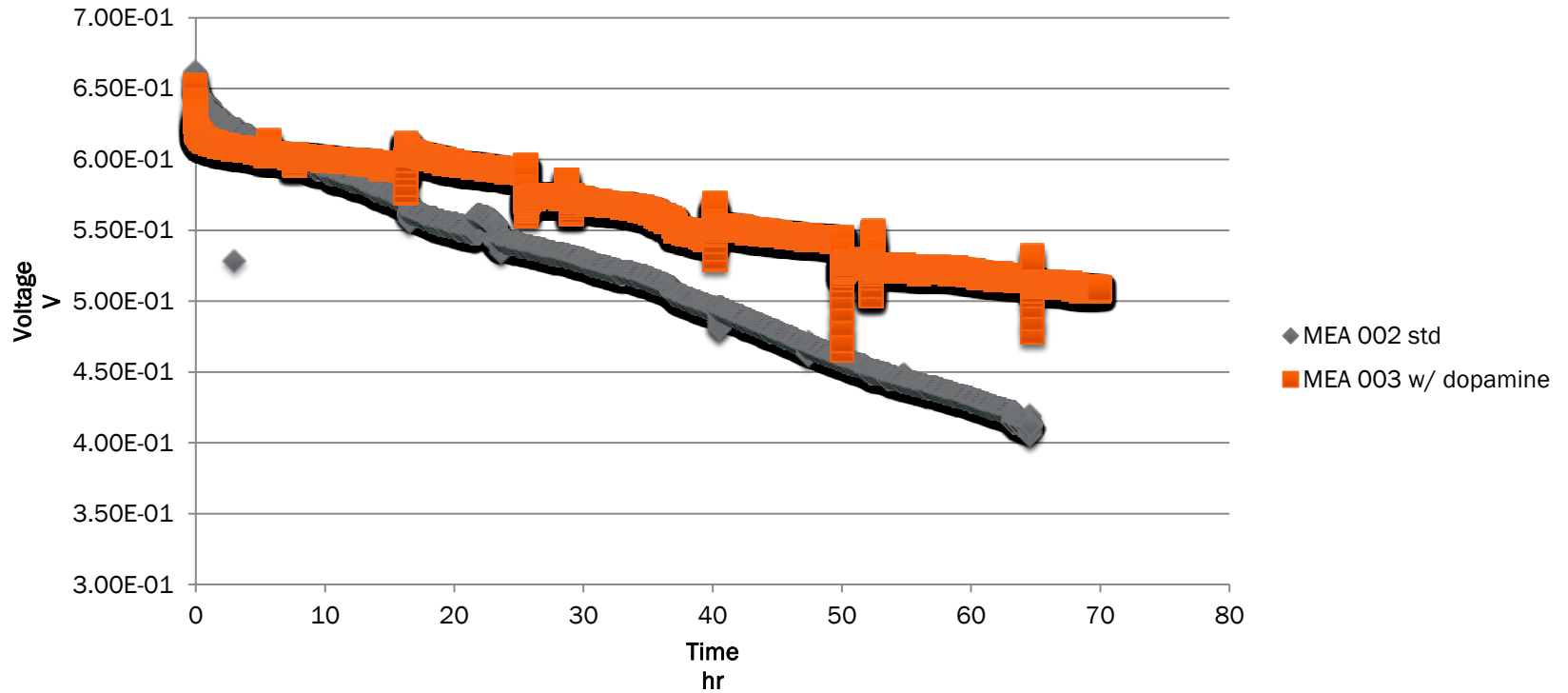
MEA002: 1.0013V

MEA003: .98165V



# LONG TERM TEST RESULTS

Constant Current at 100/90/90C 1.5atm backpressure



# FUTURE TESTS

Improve dopamine application

Improve catalyst loading

- Decrease loading

Mo<sub>2</sub>C co-catalyst incorporation

# CONCLUSION

Dopamine improves all of the key problems facing PEM fuel cells

- **Improved durability**
  - “Glue” reduces stress
  - Decreases amalgamation
  - Reduces fuel crossover
- **Decreased cost**
  - Reduced amalgamation allows for lower loading

**Preliminary tests show extremely promising results**

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# REFERENCES

(2012). Accomplishments and Progress. *Fuel Cell Technologies Office*. Retrieved March 20, 2013, from <http://www1.eere.energy.gov/hydrogenandfuelcells/accomplishments.html>

Z. Liang, T. Zhao, J. Prabhuram, A Glue Method for Fabricating Membrane Electrode Assemblies for Direct Methanol Fuel Cells, Science Direct, *Electrochimica Acta* 51 (2006) 6412–6418

V. sethuraman, J. Weidner, Hydrogen Peroxide Formation Rates in a PEMFC

Anode and Cathode: Effect of Humidity and Temperature, *Journal of The Electrochemical Society*, 155 \_1\_ B50-B57 \_2008

J. Ferrell III, M. Kuo, A. Herring, Direct dimethyl-ether proton exchange membrane fuel cells and the use of heteropolyacids in the anode catalyst layer for enhanced dimethyl ether oxidation, *Journal of Power Sources*, 195 (2010) 39–45

(2009) Procedure for Performing PEM Single Cell Testing. *Department of Energy*