

The Photocatalytic Reactions of Desert Varnish

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Outline

- What are Photocatalytic Semi Conductive Minerals?
- What is Desert Varnish?
- The process of Electron Hole Pairs.
- Experiment Results
- Implications

What are Photocatalytic Minerals?

- These are semi conductive minerals, that react with light energy and go through different redox reactions in the aqueous solution in the environment.
- They do this through a process that is known as Electron Hole Pairs
- Where can these be found?
 - Proven: Hematite and Pyrite
 - Researching: Desert Varnish

What is Desert Varnish?

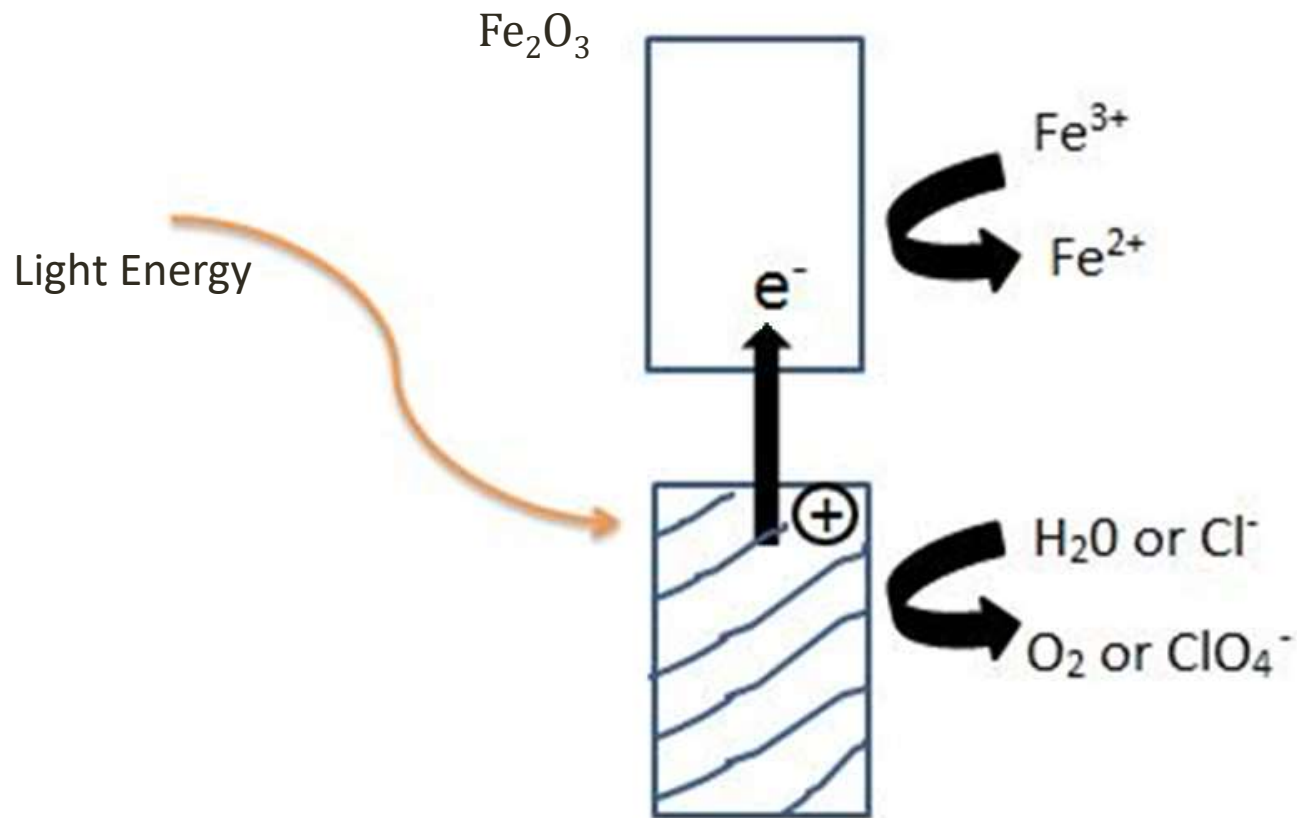
- It is a dark manganese - and iron – oxide rich coating that can be up to 100 μm thick.
- Since the 1980's it is thought to be made from Biological processes of non-phototrophic microorganisms such as cyanobacteria.



Our Hypothesis

- Desert varnish is formed by the photocatalytic properties of semiconducting minerals that occur in the varnish reacting with the aqueous solution in the environment.
- Photocatalysis creates the cations (Fe^{2+} and Mn^{2+}) that are used to stimulate and support the growth of non-phototrophic microorganisms.
- Through the process known as Electron Hole Pairs

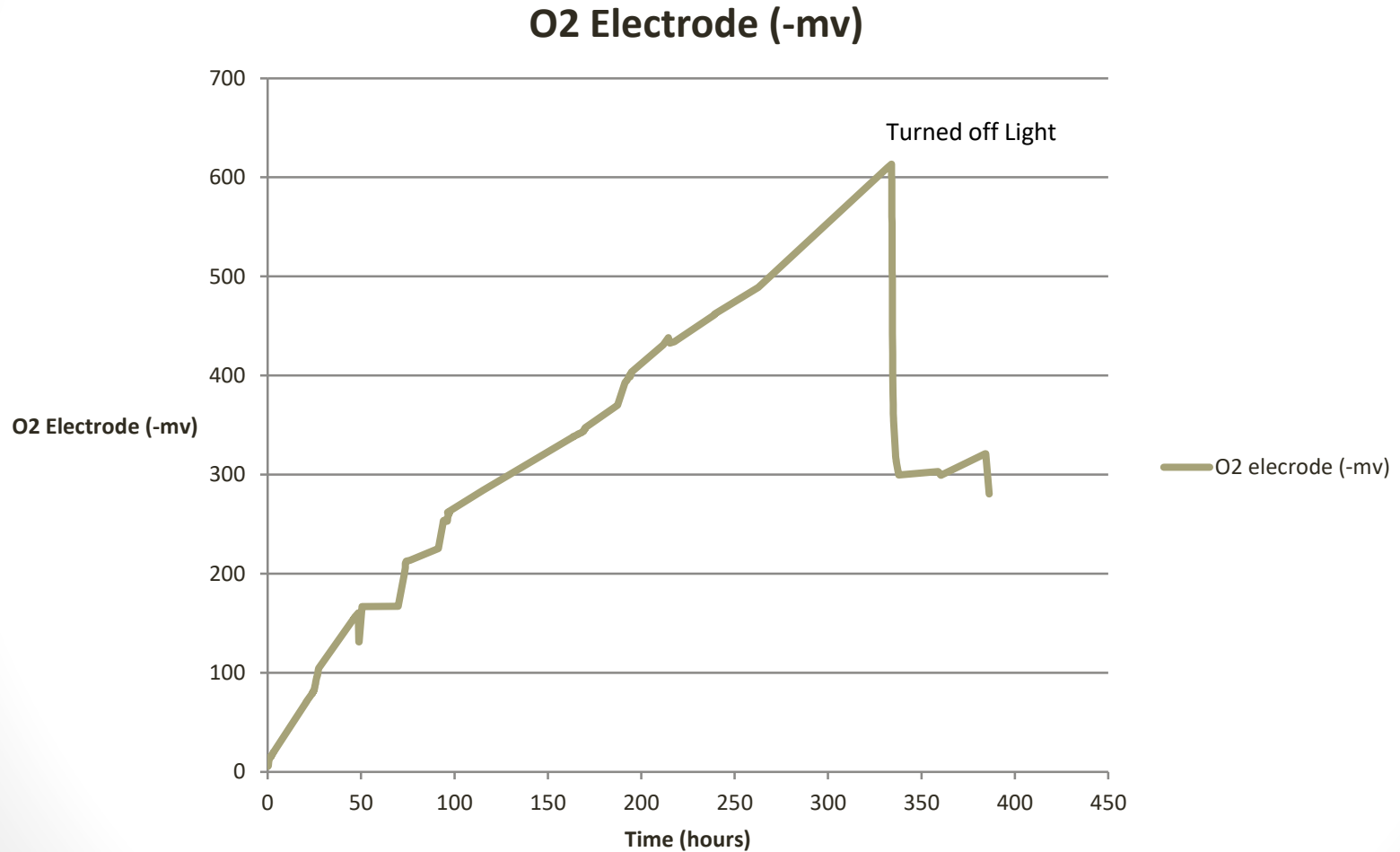
Electron Hole Pairs



Experiment

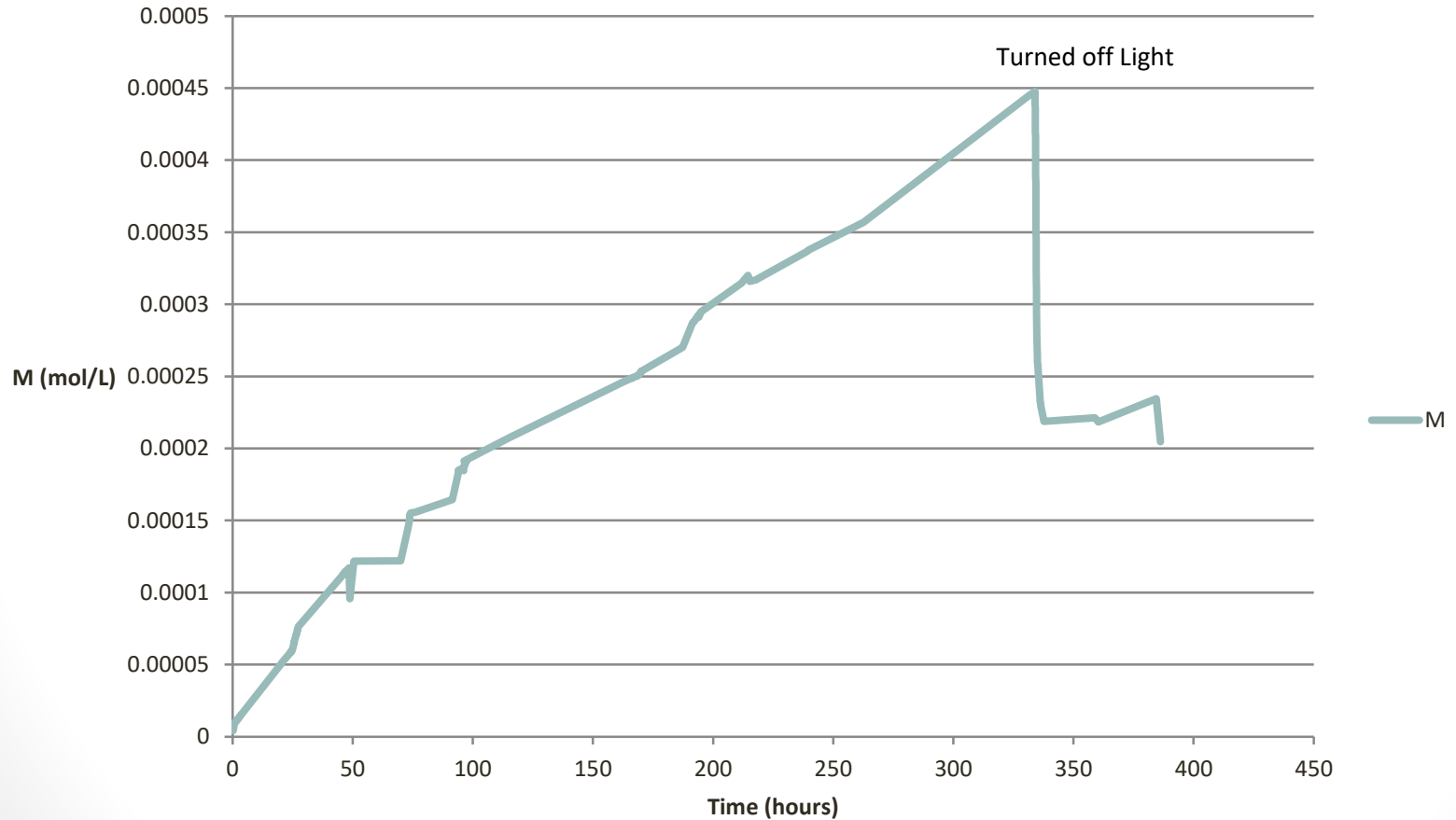
- Creating an anaerobic environment
- Placing 10 mM KCl into a air tight container with a piece of desert varnish
- Using a Xenon light source to illuminate the desert varnish in solution
- Using a dissolved free O₂ microelectrode to measure the amount of oxygen being produced in the system

Hematite and Pyrite



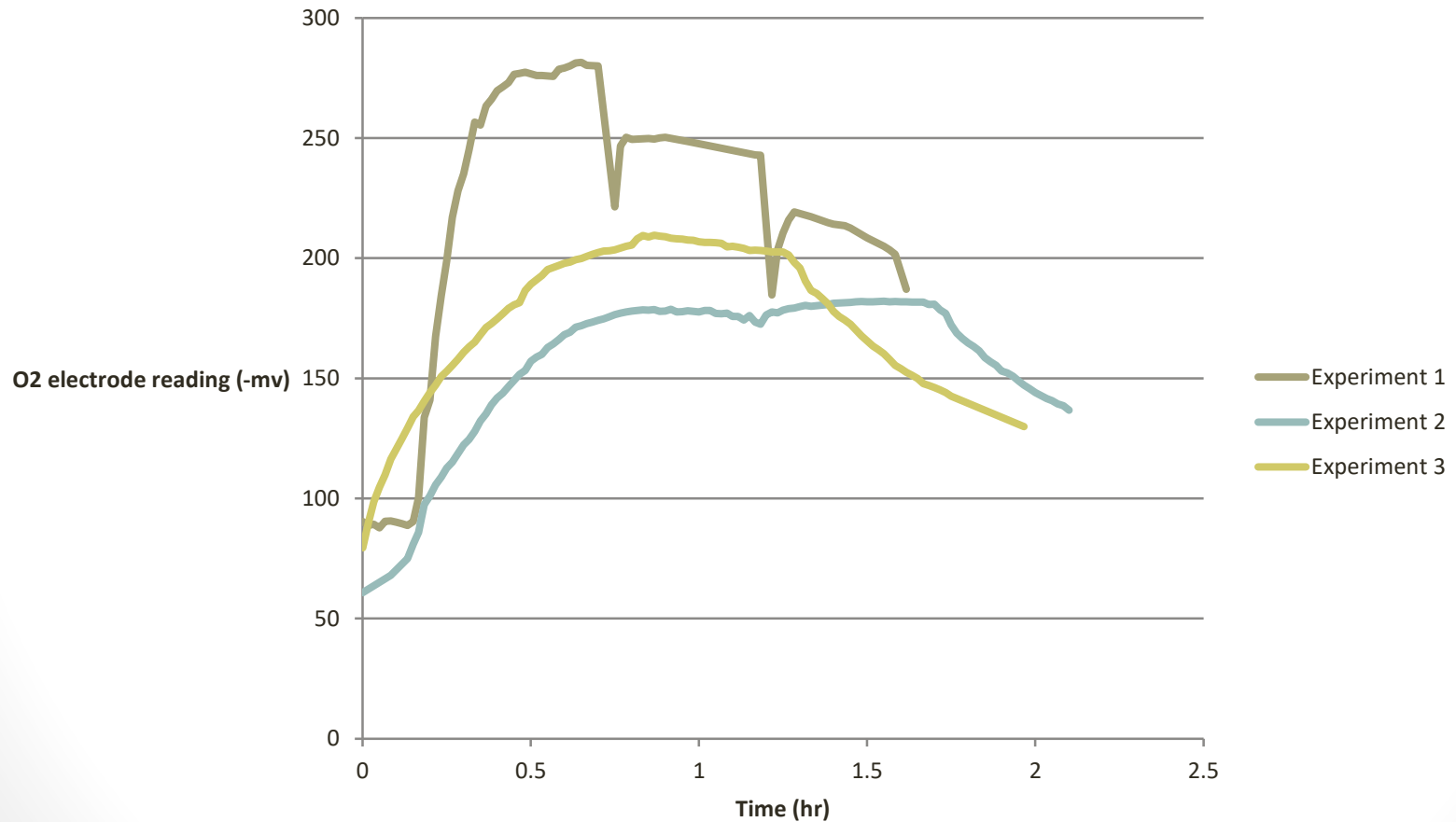
Hematite and Pyrite

M (mol/L) vs. Time



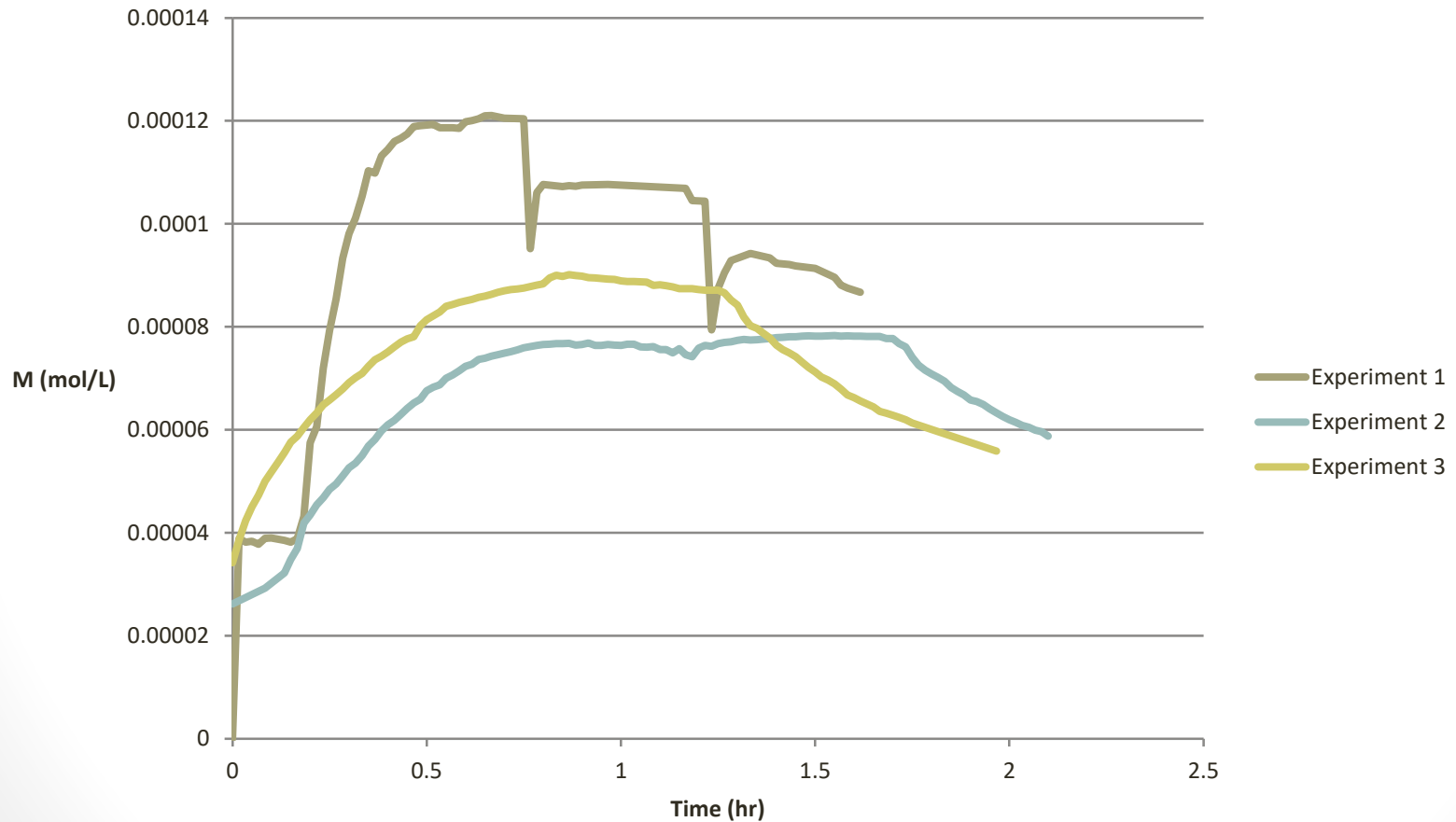
Experiments & Results

Desert Varnish O2 Electrode



Experiments & Results

Desert Varnish M Data



Implications

- Because we are creating free O₂ gas within the system containing a piece of desert varnish and 10 mM KCl we can show that the Electron Hole Pair process is occurring.
- NASA is currently using a similar process in experiments for creating a viable oxygen source on the Moon.
- If we can study this process we can possibly create solar cells from these photocatalytic semi-conducting minerals.
- There are possibilities that these reaction may also be happening on Mars which has an abundance manganese - and iron – oxide rich coatings and perchlorate.

References

- Dorn, R. I., and T. M. Oberlander. "Microbial Origin of Desert Varnish." *Science* 213.4513 (1981): 1245-247. Print.
- Eggleston, Carrick M., Justin R. Stern, Tess M. Strellis, and Bruce A. Parkinson. "A Natural Photoelectrochemical Cell for Water Splitting: Implications for Early Earth and Mars." *American Mineralogist* 97 (2012): 1804-807. Print.
- Lu, A. et al. Growth of non-phototrophic microorganisms using solar energy through mineral photocatalysis, *Nat. Commun.* 3:768 doi:10.1035/ncomms1786 (2012)
- Perry, Randall S., and Vera M. Kolb. "Biological and Organic Constituents of Desert Varnish: Review and New Hypotheses." (2003)<http://www.psi.edu:8080/about/staff/rperry/perry/SPIE_DV.pdf>.
- Potter, R. M., and G. R. Rossman. "Desert Varnish: The Importance of Clay Minerals." *Science* 196.4297 (1977): 1446-448. Print.
- Potter, R. M., and G. R. Rossman. "The Manganese- and Iron-Oxide Mineralogy of Desert Varnish." *Chemical Geology* 25 (1978): 79-94. Print.